Trivalent Ce₂O₃ and CeO_{2-x} intermediate oxides induced by laser irradiation of CeO₂ powders

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A strong non-stoichiometry of pure fcc CeO_2 was induced by laser irradiation. The increase of laser power and/or energy density had a saturable effect on particle size growth. The possibility of $CeO₂$ reduction to A- $Ce₂O₃$ by laser irradiation was demonstrated. Particles of stable $Ce₇O₁₂$ phase were observed in all specimens irradiated at low laser-power densities. An epitaxial relationship between triclinic $Ce_{11}O_{20}$ and cubic $Ce_{12}O_{22}$ phases was found. The controversial C-Ce₂O₃ phase was detected at the limits of a b c c particle. An unknown b c c phase of acicular morphology, strongly related to $C-Ce₂O₃$, was also registered. The dose dependence of $CeO₂$ structural modifications obtained by laser irradiation as a function of laser energy density variation could be explained by a simple defect aggregation model implying lattice defects (oxygen vacancies and Ce^{3+} ions).

1. **Introduction**

It is now well established that bulk cerium oxides are classified according to the oxidation state of the rareearth element as: divalent, CeO (obtained under high pressures and temperatures); trivalent, $Ce₂O₃$; and tetravalent and intermediate oxides, $CeO₂$ and CeO_{2-x} [1, 2]. Two polymorphic compounds Ce_2O_3 are also reported: $A-Ce₂O₃$ (hexagonal, space group $\overline{P3}$ ml) and the controversial C-Ce₂O₃ (cubic, space group T_h^7 or Ia3) [1, 2]. Structure and sintaxy of R_2O_3 $(R = rare-centh element)$ oxides and their relationship to the fluorite structure of $RO₂$ have been investigated in great detail [3, 4].

The intermediate oxides associated with various rare-earth elements (including cerium) can be described by the equivalent basic formulae: RO_x $(1.5 \le x \le 2)$ and R_nO_{2n-2} $(4 \le n < \infty)$ where $n = 4$ leads to R_2O_3 and $n = \infty$ to RO_2 [4-8]. The homologous series R_nO_{2n-2} contains a great number of structures (especially hexagonal, monoclinic and triclinic) built on the fluorite-like lattice of $RO₂$ compound, which are due to various oxygen-vacancy rearrangement mechanisms [3, 4, 6].

Recent electron microscopy experiments demonstrated that the electron-beam irradiation induces unexpected crystallographic phase transitions in thin films of rare-earth oxides, very sensitive to experimental conditions (especially water vapour contamination) [9, 10]. On the other hand, it is well known that laser irradiation may induce or enhance such reactions as oxidation, nitridation, reduction, and doping of various materials [11-13]. High-pressure phases of $SiO₂$ [14] and a conversion of graphite into diamond [15] were obtained by pulsed laser fluxes impinging on the respective fine powders. More recently, a laser-induced calcite-aragonite transition [16] and a reduction of V_2O_5 by the appearance of V^{4+} and V^{3+} ions in the laser irradiated powders [17], were reported.

This paper describes the various phase transformations induced by laser irradiation in pure stoichiometric $CeO₂$ powders.

2. Experimental procedure

Pure $CeO₂$ powders were prepared by precipitation of cerium nitrate of 99.991% purity using ammonia (Merck purity). After precipitation, the samples were washed for the complete removal of ammonium nitrate, then filtered and dried at 110° C for 8 h.

The irradiation of these specimens was done by a $CO₂$ laser operating in continuous wave at power densities between 0.40 and 1.30 kW cm^{-2} and for irradiation times between 7 and 50 s (Table I).

The laser beam was focused to a spot of 39 mm^2 area. The pressure of the gas mixture was 24 torr (He, 13 torr; H_2 , 9 torr; CO_2 , 2 torr). The thickness of $CeO₂$ powder layers submitted to laser irradiation was 1.5 mm.

TABLE I Experimental laser irradiation parameters

Specimen	Laser-beam power density $(kW cm^{-2})$	Irradiation time(s)	Laser energy density $(dose)^a$ $(kJ cm^{-2})$	
	0.65		4.5	
2	0.90		6.3	
3	1.00		7.0	
4	1.00	14	14.0	
5	1.15		8.0	
6	1.30		9.1	
	0.40	20	8.0	
8	0.40	50	20.0	
۹	0.40	100	40.0	

^a A conventional classification of laser irradiation doses as small $(< 7 \text{ kJ cm}^{-2})$, mean $(7-14 \text{ kJ cm}^{-2})$ and high $(>14 \text{ kJ cm}^{-2})$ doses, was used for easier discussion.

The $CeO₂$ powders were studied by conventional transmission electron microscopy (CTEM) and selected-area electron diffraction (SAED) in a TEMSCAN-100CX (Jeol) electron microscope, operating at 200 kV, with special caution in order to eliminate any significant water-vapour contamination and temperature increase during all electronmicroscopical observations.

3. Results and discussion

The pure $CeO₂$ powders consist of particles having a diameter of 20-40 nm and their structure, confirmed by SAED, belongs to the cubic polycrystalline $CeO₂$ $(a_0 = 0.541$ nm).

In general, laser irradiation induced a variable nonstoichiometry in pure fcc $CeO₂$, depending on irradiation parameters. All the specimens contained, as major phase, an fcc CeO_{2-x} (where x is very small) phase, having an increased lattice parameter $(a_0 = 0.553 - 0.555$ nm) (Fig. 1). Particles of fcc CeO_{2-x} structure ($a_0 = 0.553$ nm) were recently obtained by a vacuum annealing technique applied to CeO , [18].

A saturable effect of particle-size growth, depending on the laser power and/or energy density increase, was observed (Fig. 2). For low and mean doses $(< 9 \text{ kJ cm}^{-2})$ only CeO_{2-x} polycrystalline diffraction rings were registered, but, for higher doses, large, thin, single-crystal CeO_{2-x} platelets occurred (Fig. 3). Sometimes, two or three single-crystal particles were joined after a prolonged irradiation (Fig. 4) due to enhanced local diffusion effects.

The possibility of $CeO₂$ reduction to A-Ce₂O₃ by laser irradiation at low doses was demonstrated by SAED, because single-crystal $A-Ce₂O₃$ particles of $\langle \overline{1} 2 0 \rangle$ and $\langle \overline{4} 2 3 \rangle$ zone axis were formed in areas also containing smaller particles of CeO_{2-x} (Fig. 5). A large single-crystal A-Ce₂O₃ platelet of the same $\langle \bar{4}23 \rangle$ zone axis also occurred for mean doses (Fig. 6).

A mechanism currently used for non-stoichiometric oxide studies, based on the clustering of the simplest defect, consisting of an oxygen vacancy and two trivalent cations (virtually a $Ce₂O₃$ nucleus) resulting by laser irradiation, could explain the $A-Ce₂O₃$ appearance in a CeO_{2-x} lattice containing very high anion

Figure 1 Small fcc CeO_{2-x} particles obtained by laser irradiation at mean doses (9 kJ cm^{-2}) : (a) TEM image, (b) SAED pattern.

Figure 2 TEM images of CeO_{2-x} particles obtained by laser irradiation at: (a) low doses (4.5 kJ cm^{-2}) , (b) mean doses (14 kJ cm^{-2}) .

Figure 3 A very large single-crystal CeO_{2-x} particle which occurred in a specimen irradiated at high doses (40 kJ cm^{-2}) : (a) TEM image, (b) $\langle 343 \rangle$ zone axis pattern.

Figure 5 Single-crystal A-Ce₂O₃ particles found after laser-induced CeO₂ reduction: (a) $\langle \overline{1}20 \rangle$ zone axis pattern, (b) $\langle \overline{4}23 \rangle$ zone axis pattern (see also CeO_{2-x} diffraction rings).

Figure 4 Joined CeO_{2-x} particles after prolonged laser irradiation $(0.4 \text{ kW cm}^{-2}, 50 \text{ s}).$

vacancy concentrations. Complete conversion of $CeO₂$ to A- $Ce₂O₃$ was not possible, at least in the range of the irradiation parameters used.

The formation of stable Ce_7O_{12} phase was detected only for low laser-power densities (0.4 kW cm^{-2}) and for mean doses ($\sim 8 \text{ kJ cm}^{-2}$) (Fig. 7) or high doses (40 kJ cm^{-2}) (Fig. 8), in the last case a particular orientation of high Miller indices being identified.

Although compounds of strange non-stoichiometry of the type formed in oxidized rare-earth thin films irradiated by energetic electron beams [9, 10, 19, 20]

Figure 6 Large single-crystal A-Ce₂O₃ platelet appearing in laserirradiated CeO_{2-x} at mean doses: (a) TEM image, (b) $\langle \overline{4}23 \rangle$ zone axis pattern.

Figure 7 Particle of $Ce₇O₁₂$ phase obtained by laser irradiation at mean doses: (a) TEM image, (b) $\langle 21\overline{7} \rangle$ zone axis pattern.

Figure 8 Another Ce₇O₁₂ particle observed in a specimen irradiated at a high dose (40 kJ cm⁻²): (a) TEM image, (b) $\langle 8314 \rangle$ zone axis pattern.

were not detected, some members of the Ce_nO_{2n-2} series of greater n values were identified in the laserirradiated specimens for low power densities and short irradiation times. Thus, an epitaxial relationship between triclinic Ce₁₁O₂₀ and cubic Ce₁₂O₂₂ phases, after laser-beam irradiation of $CeO₂$ powders, was found (Fig. 9). The orientation relationships between the two phases were

$$
\quad \text{and} \quad
$$

$$
[1\bar{1}4] Ce_{11}O_{20} \parallel [\bar{1}11] Ce_{12}O_{22}
$$

 $\langle 1 1 0 \rangle$ Ce₁₁O₂₀ $\|\langle 1 1 0 \rangle$ Ce₁₂O₂₂

Figure 9 The occurrence of triclinic $Ce_{11}O_{20}$ phase from cubic $Ce₁₂O₂₂$ particles after laser-beam irradiation: (a) TEM image, (b) SAED pattern, (c) interpretation of SAED pattern (orientation relationships: $\langle 110 \rangle$ Ce₁₁O₂₀ || $\langle 110 \rangle$ Ce₁₂O₂₂ and [114] $Ce_{11}O_{20}$ [[111] $Ce_{12}O_{22}$).

The controversial $C-Ce₂O₃$ (Ia3) phase, having $a_c = 1.116$ nm and observed in some experiments [21, 22], was identified in our investigations as a thick quasiepitaxial layer grown at the limits of some prismatic particles with an unusual b c structure but having a lattice parameter $a_0 = 0.554$ nm (Fig. 10). By tilting of the $\langle 135 \rangle$ zone axis SAED pattern, the new additional spots can be interpreted as belonging to $C-Ce₂O₃$ phase having a doubled lattice parameter $(a_C \simeq 2a_o)$. A dark-field image (Fig. 10c) taken with the (635) _C spot (Fig. 10d) confirmed the C-Ce₂O₃ phase

Figure 10 (a) TEM image of a prismatic particle (b c c structure: $a_0 = 0.554$ nm), (b) the associated $\langle 135 \rangle$ zone axis pattern, (c) tilted SAED pattern with additional spots interpreted as C-Ce₂O₃ phase ($a_c \sim 2a_o$), (d) dark-field image taken with the (635)_c spot.

occurrence. It is well known that the (1 3 5) plane is the structural feature relating the homologous series in binary rare-earth oxides [4, 6]. The odd members of the Ce_nO_{2n-2} series have {1 3 5} planes that contain all the 6-coordinated rare-earth atoms with two oxygen atoms missing from each coordination cube along [111]. For even members, the defect feature is a corrugation of $\{1\,3\,5\}$ planes which result from twinning at the unit cell level. In our case, the corrugations are modified and the unit cell parameters are doubled, according to previously suggested structural principles that relate the unit cells of the members of the homologous series [5].

Another unknown bcc phase $(a_0 = 1.170 \text{ nm})$ of needle-like morphology (Fig. 11), strongly related to $C-Ce₂O₃$, was registered only in a specimen irradiated at 1 kW cm^{-2} for 7 s. The real structure of this last phase seemed to be governed by a more severe selection rule than the space group $T_h⁷$ (Ia3) because the observed lines correspond only for $h + k + l = 4n$ (Table II), excepting the very intense (2 2 2) line of the fluorite structure. This b c c phase should be an intermediate CeO_{2-x} phase having $x \gtrsim 0.5$ and an increased lattice parameter compared to $C-Ce₂O₃$ phase due to a different anion vacancy arrangement at the unit cell level.

A qualitative model could explain the observed structural modifications in fcc CeO₂ subjected to laser irradiation. At small irradiation doses, the high concentration of induced oxygen vacancies will be

Figure I1 Needle-like particles of an unknown phase, related to $C-Ce₂O₃$, found after laser irradiation (7 kJ cm⁻²): (a) TEM image, (b) SAED polycrystalline pattern.

accommodated by the open crystal structure containing multiple occupying positions for defect clusters. Because the nearest-neighbour cation sites surrounding the majority of Ce^{3+} ions will be occupied by $Ce⁴⁺$ lattice cations, the maximum extent of defect aggregation will be the formation of simple clusters involving single vacancies bound to the isolated Ce^{3+} cations. Therefore, the $A-Ce₂O₃$ occurrence could be explained by the clustering of one vacancy and two trivalent cations.

TABLE II Indexing of an unknown phase related to $C-Ce₂O₃$ phase $(T_h^7$ or Ia3)

$2R$ (mm)	d_{hkl}	hkl	I_{est} .	$I \sim pF^2d^{2a}$	$I_{\text{calc}}^{\text{b}}$
12.5	3.39	222	VS	100	100
20.6	2.06	044	S	66	33
25.4	1.65	444	m	19	8
35.5	1.19	804	w	37	12
41.5	1.02	088	W	13	6

 a p, permutation factor; F, structure factor.

b Corrected values (taking into account the angular variation of scattering atomic factors).

At mean doses, the increased number of lattice defects (oxygen vacancies and Ce^{3+} ions) will lead to an increase of anion mobility although low mobilities should be expected in the cation sublattice. Thus, a substantial ordering of interacting defects or defect complexes will occur, resulting in the subphase Ce_nO_{2n-2} (n > 6) formation [23]. In the laser-irradiated $CeO₂$ powders, we have already detected intermediate oxide phases corresponding to $n = 7$, 11 and 12.

For high irradiation doses, it might be expected that an increasing number of clusters of two, three or more $Ce³⁺$ ions would be found, thus generating a random cluster distribution in the CeO_{2-x} lattice. These trivalent ions provide deeper traps for the vacancies, resulting in a reduction of oxygen vacancy mobility. The vacancy mobility decrease could be connected to the more stable phase occurrence. Large $Ce₇O₁₂$ particles were currently observed in $CeO₂$ specimens subjected to laser irradiation at high doses.

Therefore, the occurrence of some well-known intermediate oxides after laser irradiation appears as due to the high sensitivity of the structural ordering to any small variation of oxygen vacancy concentration in the fluorite $CeO₂$ lattice.

The formation of unusual phases could be tentatively explained as a consequence of acoustic elastic waves generation by laser irradiation, The electronbeam impact has been already suggested by Gasgnier's group as being the source of high-pressure shock waves which could induce the observed paradoxical crystallography of some rare-earth intermediate oxides [9, 10, 20, 24]. Laser or electron irradiation lead to phase transition processes differing from the normal thermal annealing case, demonstrated in the case of the Jahn-Teller effect [25]. At room temperature, the induced transverse acoustic phonons result in an average displacement of the atoms from their equilibrium positions which could be approximately as large as the bond length [26]. A recent lattice dynamics calculation for high T_c superconducting oxides [27] has suggested that a coupling of acoustic waves to low-frequency oxygen modes with large amplitude of vibrations could induce phase transitions.

4. Conclusions

1. Laser irradiation induces a strong non-stoichiometry of pure stoichiometric fcc $CeO₂$. All the samples contain an fcc CeO_{2-x} having an increased lattice parameter $(a_0 = 0.553 - 0.555)$ nm) as major phase.

2. The laser-power density and/or energy-dose increase have a saturable effect on particle size growth in the CeO_{2-x} oxide powders.

3. Although the known $CeO₂$ reduction procedures are generally difficult, a new easier possibility of $CeO₂$ reduction to $A-Ce₂O₃$ by laser irradiation was demonstrated.

4. Single-crystal particles of very stable Ce_7O_{12} phase were found in all specimens laser irradiated at 0.4 kW cm^{-2} but none were detected at high power densities.

5. Although non-equilibrium compounds of strange non-stoichiometry, previously reported for rare-earth oxide thin films irradiated by electron beams, had not been observed, some unusual phases were also detected. An epitaxial relationship between triclinic $Ce_{11}O_{20}$ and cubic $Ce_{12}O_{22}$ phases was found. An unknown needle-like bcc phase and the controversial C-Ce₂O₃ phase were also registered.

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